

Iron-Sulfur Clusters in Mitochondrial Metabolism: Multifaceted Roles of a Simple Cofactor

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Abstract—Iron-sulfur metabolism is essential for cellular function and is a key process in mitochondria. In this review, we focus on the structure and assembly of mitochondrial iron-sulfur clusters and their roles in various metabolic processes that occur in mitochondria. Iron-sulfur clusters are crucial in mitochondrial respiration, in which they are required for the assembly, stability, and function of respiratory complexes I, II, and III. They also serve important functions in the citric acid cycle, DNA metabolism, and apoptosis. Whereas the identification of iron-sulfur containing proteins and their roles in numerous aspects of cellular function has been a long-standing research area, that in mitochondria is comparatively recent, and it is likely that their roles within mitochondria have been only partially revealed. We review the status of the field and provide examples of other cellular iron-sulfur proteins to highlight their multifarious roles.

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Iron is an omnipresent element on earth, and it is the second most abundant metal [1]. Iron participates in a wide range of biological cofactors and prosthetic groups. Iron can be coordinated to amino acid side chains directly (e.g. transferrins), or via porphyrin rings (e.g. hemoglobins) or sulfur atoms (such as in iron-sulfur proteins). In this review, we focus on iron-sulfur (Fe-S) proteins in mitochondrial metabolism.

The presence of Fe-S clusters (ISCs) in proteins emerged as life on earth began. It is suggested that ISCs were the first catalysts and enzyme cofactors for many biochemical reactions in the anaerobic world [2]. Their versatility and robustness can be attributed to the extraordinary properties of iron and sulfur atoms because both can readily donate or accept electrons [3–6]. There are many types of ISCs ranging from a single iron atom coordinated to four cysteine sulfhydryl groups [1Fe-0S], such

as in rubredoxin [7–9], to more complex types. ISCs such as [2Fe-2S], [3Fe-4S], and [4Fe-4S] predominate (Fig. 1). Other complex clusters also exist, such as [7Fe-8S], [8Fe-7S], or [8Fe-8S] in molybdenum-iron (MoFe) proteins [10]. In most Fe-S proteins, the cluster is coordinated to four sulfhydryl groups in cysteines. Rieske Fe-S proteins use two histidines and two cysteines instead of four cysteines for coordination [11, 12]. Other permutations of coordinating amino acids occur in some Fe-S proteins [13], and in these cases, the chemical nature of the structure is altered. Nonetheless, the basic chemical reactions in which clusters are involved remain the same.

ISCs undergo a variety of reactions and can be converted from one type to another within the course of a reaction. In Fe-S proteins, the iron atom is the donor and acceptor of electrons and it alternates between the oxidized (Fe³⁺) and reduced (Fe²⁺) states by the addition or loss of a single electron, generally without the involvement of protons [14]. *In vivo*, ISCs are incorporated into proteins by a complex pathway requiring numerous accessory proteins [15–17]. The loss or malfunction of these can lead to a wide-range of diseases (at least ten known human genetic diseases) and may ultimately cause death [18–21]. Here, we discuss the various facets of ISC biochemistry in mitochondrial metabolism.

Abbreviations: BN-PAGE, blue native polyacrylamide gel electrophoresis; cI-III, respiratory complexes I-III; CIA, cytosolic iron-sulfur protein assembly; ETC, electron transport chain; Fe-S, iron-sulfur; IRE, iron-responsive elements; IRP, iron-regulatory protein; ISCs, iron-sulfur clusters; ISP, iron-sulfur protein; mtDNA, mitochondrial DNA; Q, ubiquinone.

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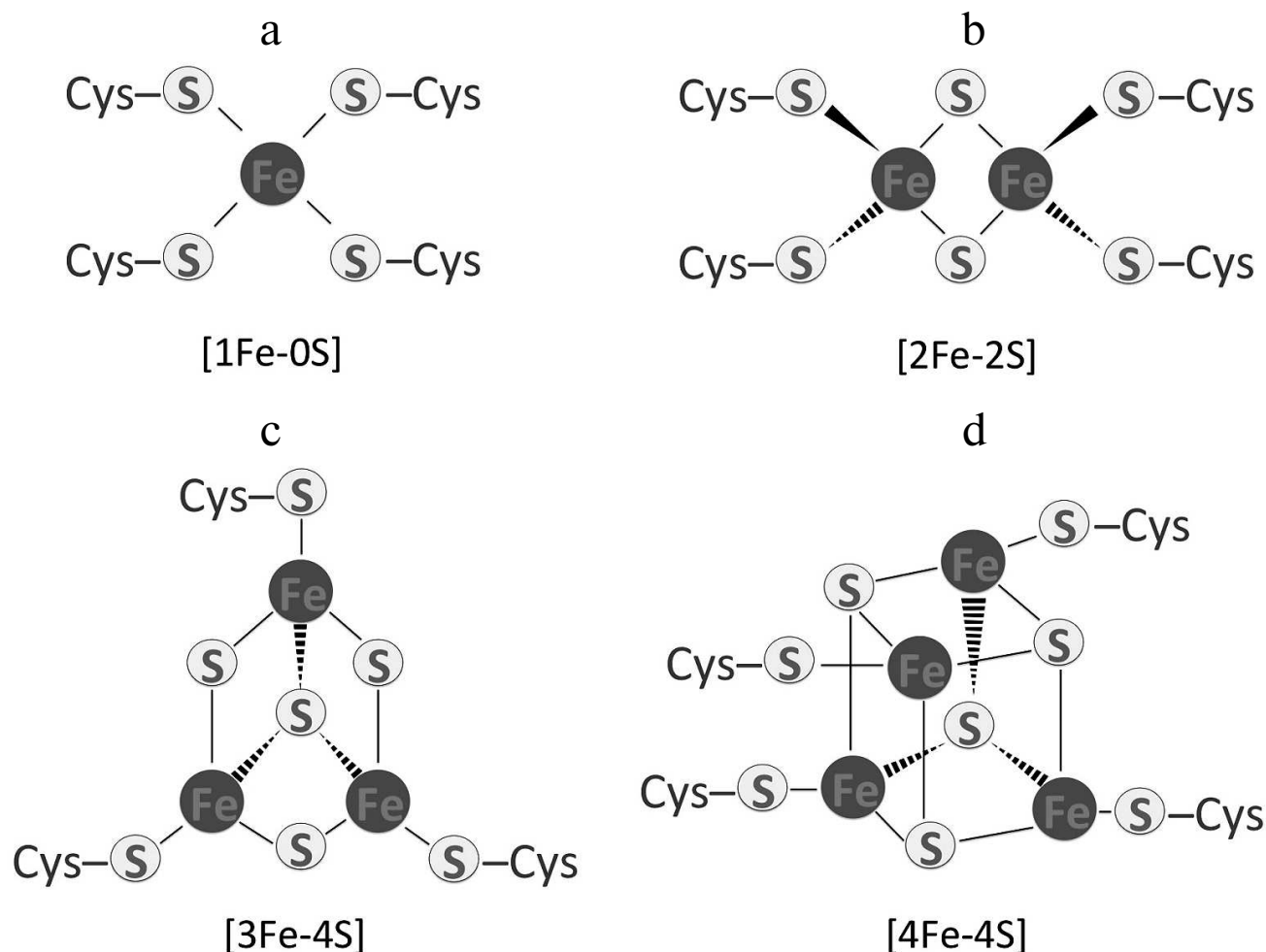


Fig. 1. Fe-S centers that are common in proteins: a) [1Fe-0S]; b) [2Fe-2S]; c) [3Fe-4S]; d) [4Fe-4S].

STRUCTURE AND ASSEMBLY OF MITOCHONDRIAL Fe-S CLUSTERS

Mitochondria share many ancestral pathways with bacteria. Among them is the Iron Sulfur Cluster (ISC) pathway of Fe-S biosynthesis [22]. Eukaryotes developed a complementary system termed the Cytosolic Iron-sulfur protein Assembly (CIA) pathway that inserts Fe-S clusters into a variety of proteins destined for the cytoplasm or nucleus [16, 23].

Structure and Biochemistry of Fe-S Proteins

ISCs occur mainly in a rhombic arrangement [2Fe-2S] or a cubane form [4Fe-4S]. The [2Fe-2S] rhombic cluster consists of two bridging sulfide ions coordinating two iron ions to four cysteines (or two cysteines and two histidines) in a protein (Fig. 1b). Recent reports indicate an unusual occurrence involving three cysteine residues and one histidine in the coordination of a [2Fe-2S] clus-

ter [24]. The [2Fe-2S] clusters occur in two oxidation states, oxidized (both irons are +3) and reduced (one iron is +2 and one is +3) [25]. In the cubane structure, four iron atoms and four inorganic sulfur atoms are coordinated to four sulfhydryl side chains of cysteines (Fig. 1d) [26]. Clusters of this kind are subdivided into two categories, low- and high-potential types. Low-potential [4Fe-4S] clusters switch between oxidized and reduced states of $[2\text{Fe}^{3+}, 2\text{Fe}^{2+}]$ and $[\text{Fe}^{3+}, 3\text{Fe}^{2+}]$, respectively. The high-potential subgroup shuttles between an oxidized state of $[3\text{Fe}^{3+}, \text{Fe}^{2+}]$ and a reduced $[2\text{Fe}^{3+}, 2\text{Fe}^{2+}]$ [5, 27].

ISCs serve a wide-range of functions (Fig. 2), and their roles can be categorized according to the chemistry involved in their reactions. Oxidation-reduction reactions and electron transfer represent the main function of ISC [28] and they serve a major role in cellular redox regulation and homeostasis. Because they have the ability to reversibly bind iron and sulfur, ISCs can also be used for Fe and S storage in the activation of specific enzymes and/or substrates (e.g. the aconitase reaction in the citric acid cycle [10, 19]). Interestingly, ISCs are also involved

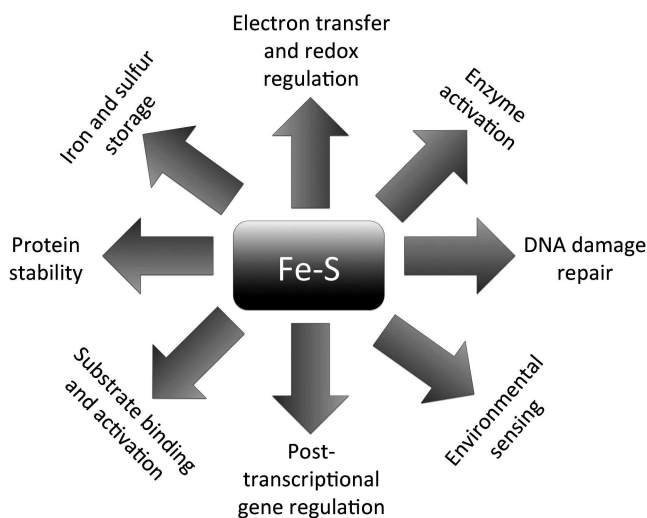


Fig. 2. Functional versatility of Fe-S clusters. ISCs in proteins serve multifaceted and unrelated functions. They are involved in redox sensing, DNA metabolism and gene activation, iron and sulfur storage, protein stability, and environmental sensing.

both directly and indirectly in regulation of gene expression [29, 30] and DNA replication [31, 32] and repair [12, 33]. Moreover, they can also enhance protein stability *in vitro* and *in vivo* [32, 34, 35].

Assembly (the ISC Machinery)

We consider here only the ISC machinery in mitochondria. Paul and Lill provide a review of the biogenesis of cytosolic and nuclear Fe-S proteins via the CIA pathway [23]. The Fe-S biogenesis pathway was first elucidated in bacteria and has been shown to be highly conserved in all species including mammals [19, 36]. Due to the complexity of and requirements for Fe-S proteins in various cellular compartments in eukaryotes, the pathway is more elaborate and requires export proteins for the crosstalk between compartments. In yeast, assembly of ISCs occurs mainly in mitochondria, whereas in mammalian cells it also occurs in the cytoplasm and nucleus [37]. Regardless of the organism, the ISC system requires a source of electrons, iron, and sulfur atoms from cysteine.

In mammalian mitochondria, Fe-S biogenesis (Fig. 3a) begins with dimerization of a cysteine desulfurase (NFS1) to form a complex with two molecules of the scaffolding protein ISCU at each end of the dimer [38]. NFS1 is stable only when bound to partner protein ISD11 [7, 39], and it is bound in the mitochondrial matrix, cytoplasm, and nucleus [8]. Both ISCU subunits bind inorganic sulfur that is provided by NFS1 in the conversion of two cysteine residues into alanine. The sulfur is then coordinated with iron that is already bound covalently by

ISCU [22, 40, 41] via linkage to cysteine residues. The source of iron is likely from the protein frataxin [13]. Frataxin has been shown to catalyze the sulfur transfer step that is rate limiting in the synthesis of [2Fe-2S] clusters [42, 43]. Frataxin binds in a groove between NFS1 and ISCU [19] in the preformed complex rather than to its individual components [44], and it induces a conformational change that activates the complex allosterically by accelerating the sulfur transfer reaction [14, 45, 46]. An electron source is needed to finalize the configuration of the nascent Fe-S structure. Some evidence indicates that ferredoxin can donate the required electrons [47]. *In vitro*, ferredoxin provides the electrons required to couple two [2Fe-2S] clusters to form a [4Fe-4S] on the ISCU scaffold protein [48]. Defects in any part of this pathway can lead to genetic diseases such as leukodystrophy and neuroregression [49], thus highlighting the important role ISCs play in mitochondrial and cellular health.

After ISC incorporation on the core complex, it can then be transferred to target proteins using co-chaperones such as mammalian mitochondrial HSC20 [50] (or bacterial HscB [51]). HSC20 interacts with a protein (containing LYR motif) that is a target ISC protein or an assembly protein for the ISC protein, forming a complex that consists of chaperone-co-chaperone-ISCU-apo-ISC protein (HSPA9-HSC20-ISCU-apo-ISC protein) [52]. Many Fe-S proteins and Fe-S protein assembly subunits in respiratory complexes II and III (cII and cIII) acquire ISCs in this manner [52, 53]. Alternatively, some of the ISCs in complex I (cI) are delivered via a different route. The mitochondrial P-loop NTPase Ind1 (an ISC assembly protein known to be required specifically for NADH dehydrogenase) transfers ISCs to apoproteins at the terminal stage in the ISC assembly process in mitochondria [54-56]. In humans, it is also called NUBPL (nucleotide-binding protein-like) [55]. Ind1 binds transiently a [4Fe-4S] cluster and transfers it to cI [54, 55, 57] (Fig. 3b). Interestingly, Ind1 shows strong specificity for cI proteins in yeasts and humans [54, 55]. The Ind1 deletion mutant in the yeast *Yarrowia lipolytica* shows only ~30% residual activity and ~20% of the relative abundance of cI compared to wild type, suggesting that the decreased activity is caused by a decrease in the cI level [54]. Likewise, a knockdown mutant of Ind1 in human HeLa cells showed a 3- to 4-fold decrease in cI activity and reduced cI assembly [55]. However, no such reduction in activity or assembly was detected in other mitochondrial Fe-S proteins such as aconitase, cII or cIII, all of which contain ISCs as cofactors [54, 55]. Despite its strong specificity for cI, the 30% residual activity of cI in the Ind1 deletion mutant suggests the possibility of involvement of other ISC delivery proteins for insertion of ISC to cI in addition to Ind1.

Ind1 may also serve other roles in mitochondria. In *Drosophila melanogaster* mitochondria, a physical interaction between a homolog of Ind1 (CG3262) and the

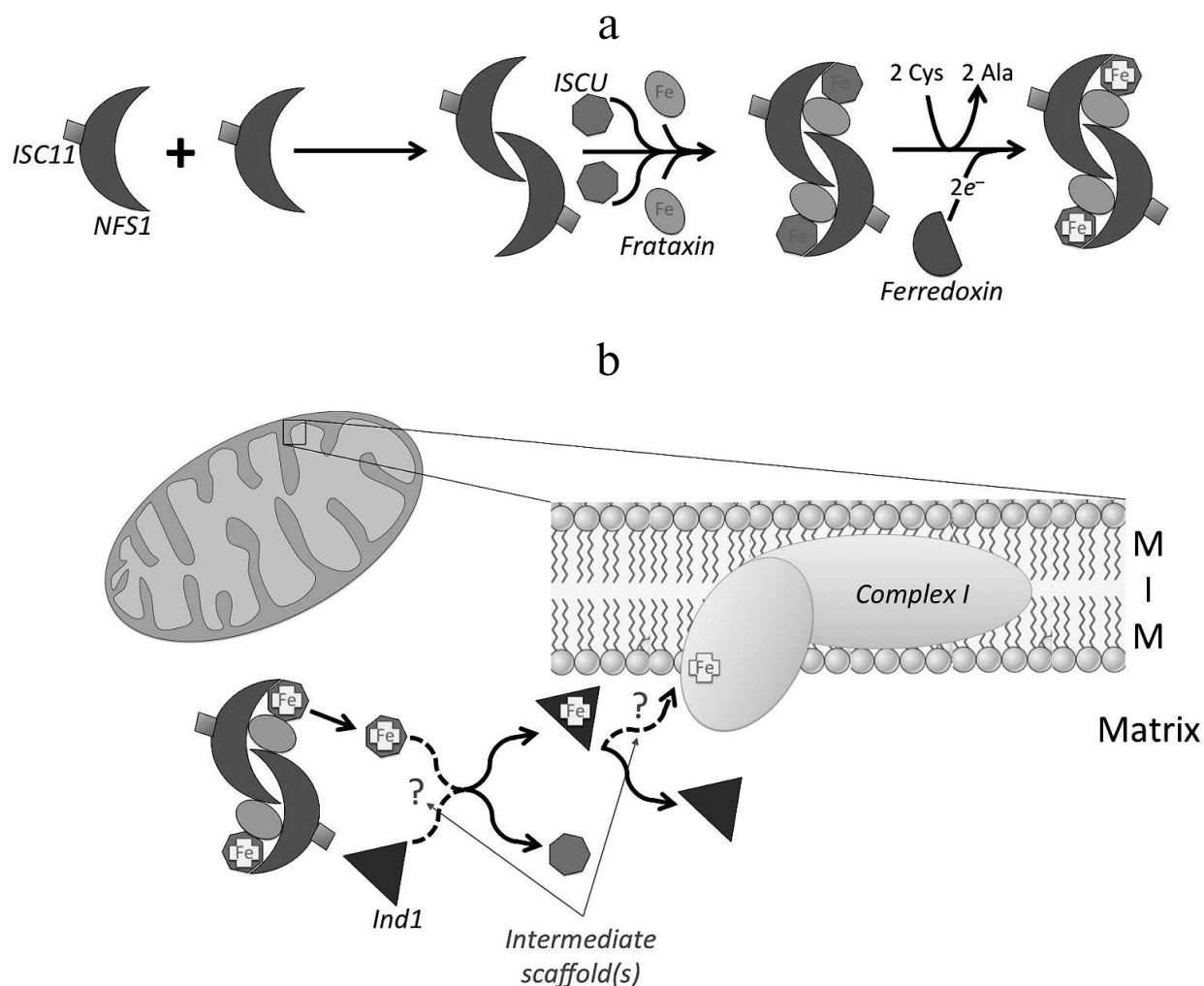


Fig. 3. Fe-S proteins are synthesized on the mitochondrial ISC machinery. a) The cysteine desulfurase protein (NFS1) in complex with the stabilizing protein ISC11 dimerizes, allowing two molecules of ISCU to bind. The binding of ISCU to both ends of the dimer creates two grooves where two frataxin molecules bind. The functional cysteine desulfurase complex extracts sulfur from cysteine, converting it to alanine. Frataxin supplies the iron, and ferredoxin donates electrons to form the ISC on ISCU. b) Ind1 is an ISC-targeting factor that has a role in the assembly of the N modules in complex I (cI) in mitochondria. It may acquire the ISC directly from ISCU, or more likely, by an indirect route involving other intermediate scaffolds. Ind1 may be targeted to the mitochondrial membrane either before or after ISC insertion, where it is involved in the transfer of [4Fe-4S] clusters to cI, either directly or again perhaps via other intermediate partners (dashed arrows and question marks) [54, 151].

mitochondrial DNA (mtDNA) replicative helicase was found by high-throughput co-affinity purification coupled with mass spectrometry [58], suggesting an expanded role for *Drosophila* Ind1 in mitochondrial DNA replication. This notion is supported by our recent discovery that the *Drosophila* mtDNA replicative helicase contains an ISC [34] (see section “Helicases”).

Fe-S CLUSTERS IN MITOCHONDRIAL RESPIRATION

Mitochondrial respiration is the main energy-yielding mechanism in aerobic eukaryotes. In oxidative phos-

phorylation, electrons donated by NADH and FADH₂ are transferred to the last electron acceptor, oxygen, passing through redox centers in four protein complexes in the mitochondrial inner membrane. Many respiratory complex proteins coordinate ISCs that are essential for their activity, and mutations in genes encoding proteins required for biogenesis of Fe-S proteins result in reduced activity of the respiratory chain [59, 60]. Electrons can be transferred directly by the reduction of Fe³⁺ to Fe²⁺ in cytochromes and Fe-S proteins. Unlike other redox centers like flavins, quinones, and other metals, both hemes and ISCs are likely to form a chain [61]. Respiratory cI (NADH:ubiquinone oxidoreductase) and cII (succinate:ubiquinone oxidoreductase) contain multiple ISCs

and transfer electrons one at a time to ubiquinone (Q) by establishing electron tunneling chains. Complex III contains only a single Fe-S protein.

Complex I (NADH:Ubiquinone Oxidoreductase)

Electrons from NADH enter the respiratory chain through the cI gateway. Complex I is a multisubunit mega protein carrying eight ISCs in five of its fourteen core subunits. The nomenclature of the eight ISCs is based on EPR signals [62, 63]: two (N1a and N1b) are [2Fe-2S] clusters, and six are [4Fe-4S] clusters (N2, N3, N4, N5, N6a, and N6b). Some prokaryotes including *Thermus thermophilus* and *Escherichia coli* contain an additional [4Fe-4S] cluster N7 in a subunit that is thought to play a role in assembly and/or structural stability rather than electron transfer due to its distal position from the main redox chain [64]. All the ISCs locate in the peripheral arm of cI, which has a hydrophilic domain and protrudes into the matrix. Ubiquinone positioned at the interface between the peripheral arm and the membrane arm [65, 66] is reduced by the N2 cluster, the terminal cluster in the cI redox chain. Complex I transfers two electrons from NADH to ubiquinone (Q) in an exergonic process that is tightly coupled to the endergonic translocation of four protons across the membrane into the acidic intermembrane space [67]. The ISCs in cI are described below; individual proteins are indicated by their human/bovine designations.

N3: NDUFV1/51 kDa. Cluster N3 is a [4Fe-4S] cluster that is positioned within ~ 8 Å of flavin mononucleotide (FMN) in subunit NDUFV1/51 kDa, and is the first ISC in the redox chain. One electron from FMN is transferred by N3 to N1b in NDUFS1/75 kDa via intersubunit transfer [68, 69].

N1b, N4, N5: NDUFS1/75 kDa. NDUFS1/75 kDa carries three ISCs; N1b is a [2Fe-2S] cluster and N4 and N5 are [4Fe-4S] clusters [70]. NDUFS1/75 kDa transfers an electron from N1b to N4 to N5. Notably, N5 is coordinated by three cysteines and one histidine, so its EPR properties differ from clusters coordinated by four cysteine residues [70]. The longest edge-to-edge distance between ISCs is that between N5 in NDUFS1/75 kDa and N6a in NDUFS8/TYKY. Thus, the electron transfer rate between N5 and N6a is the rate-limiting step in the electron transfer pathway within the ISC chain in cI. Nonetheless, molecular dynamic simulations suggest that a water molecule in the intersubunit space can enhance the rate of transfer by near three orders of magnitude [71].

N6a, N6b: NDUFS8/TYKT. NDUFS8/TYKT has two [4Fe-4S] clusters and transfers an electron from N6a to N6b. N6a is located near the interface of the N module and the Q module of cI near the zinc-binding site of a cI accessory subunit, NUMM (in *Y. lipolytica*) or

NDUFS6 (in humans) [72]. Mutations in Zn-coordinating residues of NUMM compromise proper assembly of cI, and deletion of NUMM causes reduction in the EPR signal of N6a, suggesting that stable insertion of N6a requires the Zn-binding site in NUMM [72].

N2: NDUFS7/PSST. Cluster N2 is coordinated in NDUFS7 near its interface with NDUFS8. Due to its higher redox midpoint potential, N2 receives electrons from other clusters as an electron sink [70] and transfers them to ubiquinone exiting the electron transport chain at cI [73]. The midpoint potential of N2 shows a pH dependence, becoming more positive at lower pH values [74]. This pH dependence is due to the protonated group, His226 of the 49 kDa (NDUFS2) subunit in the case of *Y. lipolytica*. Mutation of this histidine abolishes the pH dependence, but surprisingly does not affect the proton pumping mechanism, implying that the ISC redox chain is not linked directly to proton pumping. Rather, it is suggested that the interaction between N2 and semiquinone species may be linked to a redox-driven coupling mechanism [73, 75, 76].

N1a: NDUFV2/24 kDa. NDUFV2/24 kDa carries a [2Fe-2S] cluster, N1a, in a hydrophobic surrounding. N1a is not part of the main electron redox chain; its position is too remote and it has a very low mid-point potential. Because the mid-point potential is higher than that of flavosemiquinone, it is thought that one of two electrons from FMN is transferred to N1a and the other to N3 [63]. It has also been suggested that N1a serves a role in preventing ROS production [77]. Though its specific role has not been elucidated, N1a is conserved across species, suggesting its importance.

Complex II (Succinate:Ubiquinone Oxidoreductase; Succinate Dehydrogenase (SDH))

Complex II comprises four nuclear-encoded polypeptides; a flavoprotein (SDHA) and a Fe-S protein (SDHB) form a hydrophilic head in the matrix and are tethered to a membrane anchor domain that consists of SDHC and SDHD [78]. Complex II transfers electrons derived from the oxidation of succinate to fumarate in the citric acid cycle via FADH₂ to ultimately reduce the mobile electron carrier ubiquinone, coupling the citric acid cycle and the electron transport chain (ETC). Complex II (SDH) is the only component of the citric acid cycle that is membrane-bound. It is distinguished from other complexes in the ETC by its inability to pump protons, in addition to its lack of any subunit that is encoded by mitochondrial DNA. The two reactions catalyzed by cII, the oxidation of succinate in the flavoprotein (SDHA) and the reduction of ubiquinone in the membrane-anchored domain (SDHC + SDHD), are linked via electron transport through three ISCs in SDHB [79].

Succinate dehydrogenase B (SDHB). SDHB contains three different ISCs: [2Fe-2S], [4Fe-4S], and [3Fe-4S]. The rhombic [2Fe-2S] cluster is ligated by four cysteine residues in the N-terminal domain comprising an α -helix and five β -strands. It is located adjacent to FAD in the flavoprotein (SDHA). Both [4Fe-4S] and [3Fe-4S] are in the C-terminal domain, which contains six α -helices that interact largely with the membrane anchor domain. The three ISCs are aligned almost linearly. Each edge-to-edge distance is less than 14 Å, indicating a favorable electron transfer [61]. The [3Fe-4S] cluster is in the ISC chain in SDHB and lies 7.1 Å away from ubiquinone and 13.3 Å from heme [61, 78]. Thus, reduction of ubiquinone occurs prior to that of hemes, as is anticipated from their respective redox potentials [78].

Insertion of the ISCs into apo-SDHB occurs in the mitochondrial matrix prior to formation of a heterodimer with SDHA; it is guided by HSC20, a co-chaperone in the ISC biogenesis pathway [80]. Rouault and coworkers [52, 53] showed in a yeast two-hybrid screen that SDHB has three independent binding sites for HSC20; two sites have a L(I)YR motif and one has a KK_{x6-10}KK motif. These are the two most prevalent consensus sequences for HSC20 binding and are found in Fe-S cluster recipient proteins [52, 53].

SDHAF1, an assembly protein for cII, also contains a L(I)YR motif, implying that it may enable the insertion of ISCs into SDHB, using its interaction with HSC20 [52]. Co-immunoprecipitation analysis and mitochondrial subfractionation coupled with Blue Native Polyacrylamide Gel Electrophoresis (BN-PAGE) results show that three clusters are transferred by a chaperone/co-chaperone system through either the HSC20–HSPA9–ISCU–SDHB complex or the HSC20–HSPA9–ISCU–SDHB–SDHAF1 complex [52]. As expected, mutations in SDHAF1 can cause succinate dehydrogenase deficiency manifesting as infantile leukoencephalopathy with accumulation of blood succinate and lactate [21].

Complex III (Ubiquinol:Cytochrome *c*-Oxidoreductase)

Electrons are shuttled from cI and/or cII via ubiquinol (the reduced form of ubiquinone) to cIII. Complex III transfers an electron from ubiquinol to cytochrome *c* and recycles the other electron from ubiquinol for proton motive force generation through the Q-cycle mechanism in a bifurcated fashion. Structural studies demonstrate that cIII is a dimer [81-83] that contains three essential redox subunits: cytochrome *b*, cytochrome *c*₁, and the Fe-S protein (ISP), although the total number of subunits varies from three in prokaryotes to eleven in humans [84]. It harbors three types of redox centers: two *b*-type hemes, one *c*-type heme, and a [2Fe-2S] cluster. ISP mediates one electron transfer from ubiquinol to cytochrome *c*₁ mod-

ulated by another electron transfer pathway through cytochrome *b* [84-86].

Iron sulfur protein (ISP). ISP in cIII is also called “Rieske” protein because it has a Rieske-type [2Fe-2S] cluster that is coordinated by two cysteines and two histidines [82, 85-88]. ISP is anchored to the mitochondrial inner membrane by a transmembrane helix with a soluble extramembrane domain called the extrinsic domain at its C-terminus [81, 82, 89]. The extrinsic domain (ISP-ED) harbors the [2Fe-2S] cluster in the intermembrane space. This domain moves to the Q_o site or to cytochrome *c*₁ depending on the occupancy of the Q_o site by inhibitors, enabling the [2Fe-2S] cluster to accept an electron from the site of oxidation of ubiquinol and to donate the electron to an oxidized cytochrome *c*₁, respectively [81, 82, 90]. Although it is clear that a substantial conformational change of the ISP-ED is necessary for the bifurcated electron transfer to overcome the >14 Å distance and unfavorable rate, the detailed mechanism is still unknown. Two models have been proposed to explain movement: (i) the binding affinity modulated ISP-ED motion switch model [84, 90] and (ii) the two-position model [91]. Interestingly, a mutation in cytochrome *b* of *Rhodobacter capsulatus* (G167P) shifts the movement of ISP-ED toward positions far from the Q_o site and induces the production of superoxide radicals [92]. A corresponding mutation in the human enzyme (S151P) has been identified as a mitochondrial disease-related mutation [92]. Gurung et al. reported that cIII lacking its ISC creates a proton leak, suggesting a role for the [2Fe-2S] cluster in gating a proton channel [86].

Insertion of the [2Fe-2S] cluster in ISP is assisted by LYRM7, an assembly factor shown to bind to it in human cIII [93]. LYRM7 is co-immunoprecipitated with HSC20. That the level of ISP is reduced after knockdown of HSC20 suggests LYRM7 also participates with the HSC20–HSPA9–ISCU complex for ISP assembly in cIII, as does SDHAF in assembly of cII [52].

Fe-S CLUSTERS IN MITOCHONDRIAL METABOLISM

Fe-S clusters play important roles in the various metabolic pathways within mitochondria. We discuss below the recent literature on Fe-S involvement in mitochondrial metabolism.

Citric Acid Cycle

The citric acid cycle is the central cyclic pathway by which the carbon skeleton of glucose is released as carbon dioxide, producing energy in the form of ATP and reduced coenzymes [94]. A molecule of oxaloacetate is regenerated in a sequence of eight enzyme-catalyzed reactions. Three enzymes contain ISCs: aconitase (which

can be considered as a moonlighting enzyme with dual roles involving its ISC), succinate dehydrogenase (common to both the citric acid cycle and the electron transport chain, see section "Fe-S Clusters in Mitochondrial Respiration"), and fumarase (in several bacteria and a single archaeon, to date).

Aconitase. Aconitase, the second enzyme of the citric acid cycle, catalyzes the conversion of citrate to isocitrate via an alkene intermediate (*cis*-aconitate). It has two conformations depending on its activity, both of which coordinate an ISC. The inactive protein comprises four domains and binds a [3Fe-4S] cluster [95]. Once activated, it acquires another iron atom to form a [4Fe-4S] center [96, 97]. The cluster is coordinated by three cysteine residues, and upon insertion, its labile iron is coordinated by water molecules. The additional iron atom is crucial for the activity of the enzyme and can be removed by a variety of oxidants. For example, thiocyanate (an oxidant elevated in the arteries of smokers) releases it both in the isolated enzyme and in cultured cells, leading to protein dysfunction [98]. Chromate ions also cause the oxidative inactivation of aconitase and other Fe-S proteins [99]. Similarly, low doses of the strong oxidant peroxyxynitrite release the labile iron [100].

A cytosolic isozyme of mitochondrial aconitase termed iron-regulatory protein (IRP) or iron-responsive element binding protein (IRE-BP) [101] has aconitase activity and serves another role in cellular iron homeostasis, a function that is also mediated through its ISC [102]. It is an mRNA-binding protein that binds iron responsive elements (IRE) in mRNA, resulting in either its stabilization or degradation, thus providing a posttranscriptional control point. The key players in cellular iron homeostasis are ferritin (an iron-storage protein) and the transferrin receptor (a membrane gateway for the cellular entry of iron from blood). Apo-aconitase (carrying a [3Fe-4S] cluster because of low cytosolic iron concentration) binds and stabilizes the mRNA for the transferrin receptor, promoting protein production and increasing iron transport into the cytoplasm. At the same time, apo-aconitase binds to the ferritin mRNA and prevents its translation under cellular conditions of low iron [94]. When iron levels are elevated in the cytoplasm, apo-aconitase acquires the labile iron and induces a conformational change to holo-aconitase carrying a [4Fe-4S] cluster, leading to the availability of free ferritin mRNA and subsequent ferritin production [103]. As with the mitochondrial isoform, cytosolic aconitase is also regulated through its ISC [104]. Aconitase activity is inhibited in cells treated with thiocyanate resulting in increased IRP-1 activity and ultimately higher levels of iron in the cytoplasm, possibly resulting in toxic effects [98]. Interestingly, mitochondrial aconitase in yeast has also been shown to serve a role in mitochondrial DNA (mtDNA) maintenance independent of its catalytic activity [105]. Whether mitochondrial or cytosolic, aconitase presents a case of dual function mediated by its ISC.

DNA Metabolism

Until recently, knowledge of the presence of Fe-S clusters in enzymes specialized in nucleic acid metabolism was rare [106]. Now numerous enzymes involved in DNA and RNA transactions have been shown to carry one or more ISCs of various types and structures. The presence of ISCs in primases, helicases, nucleases, ligases, glycosylases, polymerases, and transcription factors has proved to be essential for protein structure and function [107]. To date, only a few proteins in mtDNA metabolism have been demonstrated to contain ISCs, but this number is highly likely to become substantially larger. ISCs function as a typical redox-sensing, electron-transferring cofactor in some nucleic acid processing proteins. For example, the EndoIII and MutY DNA repair glycosylases sense DNA damage via charge transfer within their Fe-S clusters [108, 109]. Some RNA metabolism enzymes use ISCs to transfer electrons to S-adenosylmethionine to mediate methylation of target rRNAs and tRNAs [110]. In addition, many transcription factors that sense nitric oxide regulate gene expression via small changes in their ISCs [4]. ISCs also serve multifarious other roles. We present below a short description of current knowledge; the table summarizes recent findings about some ISC-carrying enzymes in DNA metabolism, including the several that have been identified in mitochondria.

Helicases. DNA helicases play a central role in nucleic acid metabolism to unwind double-stranded DNA using the energy provided by nucleotide hydrolysis to translocate along and provide access to single-stranded DNA in the process of transcription, replication, recombination, and repair [122].

ISCs in helicases have been identified largely in DNA helicase superfamilies 1 and 2 (SF1 and SF2) [123]. An exception is the mitochondrial replicative DNA helicase, which is a SF4 helicase [34, 124]. The XPD/FANCI family including XPD, FancJ, Chl1, and RTEL has been studied extensively due to its strong association with human diseases [125]. XPD was the first ISC-carrying DNA repair helicase in SF2 to be identified [126]. Its [4Fe-4S] cluster is located in the core helicase domain. Both structural [127, 128] and biochemical [129] analyses show that the ISC in XPD is proximal to the duplex DNA strand separation site, suggesting a role for the ISC in determination of the unwinding point. ISCs in XPD and FancJ are also essential for helicase activity in nucleotide excision repair [112, 126]. Consequently, mutations in human XPD cause several genetic diseases such as *Xeroderma pigmentosum*, Cockayne syndrome, Fanconi anemia, and trichothiodystrophy [112, 126]. Furthermore, Fe-S-cluster DNA helicases were shown to be inhibited potently by protein-DNA interactions, thereby affecting DNA metabolism. For example, the helicase activity of FancJ is inhibited by shelterin proteins that bind telomeres [130].

Structure and function of ISCs in some DNA metabolism enzymes

Protein	Type	Organism	Subcellular location	ISC type	Known function(s)	References
Replicative helicases	mtDNA helicase	<i>Dm</i> (E)	Mito	[2Fe-2S]	stability	[34]
Repair helicases	XPD	<i>St</i> (A)	Nuc	[4Fe-4S]	helicase activity	[111]
	FancJ	<i>Hs</i> (E)	Nuc	[4Fe-4S]	helicase activity	[112]
	Rtel1	<i>Hs</i> (E)	Nuc	[4Fe-4S]	redox signaling	[113]
	ChlR1/DDX11	<i>Hs</i> (E)	Nuc	[4Fe-4S]	unknown	[114]
Helicase-nucleases	AddAB	<i>Bs</i> (B)		[4Fe-4S]	stability, DNA binding	[115]
	Dna2	<i>Sc</i> (E)	Nuc and Mito	[4Fe-4S]	helicase and nuclease activities and crosstalk	[35, 116]
Nucleases	Exonuclease 5	<i>Sc</i> (E) <i>Hs</i> (E)	Mito and Nuc	[4Fe-4S]	exonuclease activity	[117, 118]
	Cas4	<i>Ss</i> (A)		[4Fe-4S]	DNA binding, nuclease activity	[119, 120]
	Cas4	<i>Pc</i> (A)		[2Fe-2S]	no effect on nuclease activity	[121]

Notes: A, Archaea; B, Bacteria; E, Eukarya; *Bs*, *Bacillus subtilis*; *Pc*, *Pyrobaculum calidifontis*; *Ss*, *Sulfolobus solfataricus*; *St*, *Sulfolobus tokodaii*; *Sc*, *Saccharomyces cerevisiae*; *Dm*, *Drosophila melanogaster*; *Hs*, *Homo sapiens*; Mito, mitochondria; Nuc, nucleus.

Mitochondrial replicative DNA helicase. The mitochondrial replicative DNA helicase from *Drosophila melanogaster* is the only known ISC-containing replisome protein identified in mitochondria to date. Our group has shown recently that its N-terminal domain contains a [2Fe-2S] cluster that is essential for protein stability *in vitro* [34]. Its ISC is bound by the homologous cysteine residues that coordinate zinc in the primase-helicase from bacteriophage T7. This evolutionary switch from zinc to iron binding is intriguing, particularly because it resides in the N-terminal part of the protein that has not yet been ascribed a clear function.

Helicase-nucleases. Nucleases cleave phosphodiester bonds by various catalytic mechanisms [131]. Some nucleases have evolved into bifunctional enzymes carrying two catalytic domains in a single polypeptide: a helicase domain for unwinding DNA and a nuclease domain for phosphodiester bond cleavage [132]. Two helicase-nucleases in the RecB family have been shown to carry ISCs: AddAB and Dna2. AddAB is a well-studied Fe-S helicase-nuclease from the bacterium *Bacillus subtilis* [133]; it contains a 4Fe-4S cluster in its C-terminal domain with roles in maintaining the structural stability of the nuclease domain and binding to DNA ends [115]. Similar structures and roles have been found in other nucleases outside the helicase-nuclease group, including

yeast exonuclease 5 [117] and the archaeal nuclease Cas4 from *Sulfolobus solfataricus* [119].

Dna2. Dna2, a helicase-nuclease in the RecB family, was identified as an ISC-containing enzyme that localizes to both the nucleus and mitochondria [35, 134, 135]. Its ISC was predicted based on the presence of a conserved ISC signature motif within the nuclease active site [116]. Pokharel and Campbell demonstrated the presence of the ISC in the Dna2 helicase-nuclease from *Saccharomyces cerevisiae* by UV-vis spectrometry and mutagenesis of the ISC-ligating cysteines, which causes reduction in both its nuclease and ATPase activities, suggesting a possible role of the ISC in linking its nuclease and helicase functions [35]. However, variants lacking the cluster showed no significant difference in DNA binding [35]. Dna2 is involved in maintaining the integrity of both the nuclear and mitochondrial genomes [134, 135]. Its role in yeast nuclear DNA replication and repair in Okazaki fragment processing and double-strand break repair, respectively, are well-characterized [136]. Zheng et al. demonstrated that human Dna2 stimulates mtDNA polymerase activity and is involved in removal of RNA primers and intermediates of long-patch base excision repair in mitochondria [134].

Nucleases. The structural signature of the nuclease domain in which the ISC is bound in the AddAB heli-

case-nuclease is also found in other nucleases including exonuclease 5 and Cas4 [107, 115]. An *in vitro* study showed that the ISC of the Cas4 nuclease from *S. solfataricus* serves a role in protein stability [119]. The crystal structure of Cas4 revealed a toroidal form comprising five dimers, with each protomer containing a 4Fe-4S cluster that is required for DNA binding and cleavage activity [120]. Interestingly, Cas4 from *Pyrobaculum calidifontis* carries a 2Fe-2S cluster, indicating species-specific differences. However, the loss of the ISC apparently has no effect on nuclease activity [121].

Yeast exonuclease 5. Yeast exonuclease 5 (EXO5) localizes to mitochondria and serves an essential role in mitochondrial genome stability [117]. The presence of an ISC was predicted by sequence alignment with RecB family nucleases, and it shares conserved cysteine residues with those in the nuclease domain of AddAB [117]. The presence of an ISC was subsequently demonstrated by UV-vis spectrometry in its human homolog, hEXO5 [118]. A mutant lacking the ISC in hEXO5 exhibits only ~10–20% residual nuclease activity; however, it may function only in nuclear DNA repair because unlike yeast EXO5, hEXO5 gene lacks a mitochondrial leader sequence [118].

Apoptosis

Mitochondria are not only essential for the vitality of cells; they serve a strategic role in committing cells to undergo programmed cell death [137]. Intrinsic apoptosis is mediated by the intactness of the mitochondrial outer membrane. A breach in the outer membrane results in the release of proteins from the intermembrane space (specifically cytochrome *c*), and a cascade of events eventually leads to apoptotic cell death [138, 139]. Fe-S proteins play multifarious roles in mitochondrial iron homeostasis and apoptosis. MitoNEET is an integral mitochondrial outer membrane protein that has been implicated in iron homeostasis in both mitochondria and the cytoplasm [11, 140, 141]. It contains a redox-active [2Fe-2S] cluster [24, 142]. The induction of cell death through treatment with TNF α is mediated by the binding of the Stat3–Grim–19 complex to mitoNEET, forcing the rapid release of its ISC. Mitochondria then accumulate iron in the matrix, inducing the formation of reactive oxygen species and mitochondrial injury accompanied by cell death [143]. MitoNEET has been shown recently to act as an ISC transfer protein [144] that uses a redox switch mechanism to regulate transfer [145]; its redox-sensing role helps to combat oxidative injury by recovery of labile Fe-S proteins [146]. In pancreatic β cells, overexpression of mitoNEET leads to an increased sensitivity to TNF α cytotoxicity, and the effects of TNF α can be attenuated if it is unable to transfer its ISC (via pharmacological intervention) [147]. MitoMEET has also been implicated in

the proliferation of human breast cancer cells and promotion of tumor growth [148].

Cancer can also be induced by the absence of p53, the “guardian of the genome”. Lack of p53 in tumor cells was shown recently to decrease utilization of mitochondrial iron by downregulation of the expression of human frataxin [149]. Cellular iron deficiency is also detrimental for lipid synthesis and will eventually alter the properties and functions of biological membranes [150], thereby affecting multiple mitochondrial pathways.

The CIA machinery for ISC assembly in the cytoplasm is also involved in mitochondrial integrity and/or apoptosis. Human cytokine induced apoptosis inhibitor (CIAPIN1) belongs to the anamorsin protein family and has been shown to function as an antiapoptotic protein through the regulation of Bcl-2 and Bax [151]. Anamorsin carries a [2Fe-2S] cluster and functions early in the CIA pathway as part of an electron transfer chain [152]. The yeast homolog of anamorsin, Dre2, is also a Fe-S protein to which electrons are transferred from Tah18 in the early steps of CIA [153]. The interaction of Tah18 and Dre2 is lost when yeast cells are exposed to lethal doses of H₂O₂ and Tah18 is relocalized to mitochondria. There it mediates the loss of outer membrane integrity, promoting apoptosis. Interestingly, anamorsin was able to substitute for yeast Dre2 and in its interaction with Tah18 [154].

CONCLUSIONS AND FUTURE DIRECTIONS

The requirements for and versatility of iron-sulfur clusters in myriad metabolic pathways in organisms across taxa has led to intensive research on their structural and functional properties. The roles of ISCs in various cellular compartments and the influence of environmental factors upon them have been explored in detail in multiple systems. In mitochondria, their function in respiration is well documented, as is the pathway of their synthesis and transfer to recipient proteins. Other areas of mitochondrial Fe-S biology are less well explored, and the function of ISCs in mitochondrial nucleic acid metabolism remains somewhat enigmatic. To our knowledge, only three such proteins have been identified to date, each playing an important role in DNA replication and/or repair, and perhaps recombination, but their evolutionary histories are not readily apparent; each of them may not be present in all animal taxa and/or they may not all carry an ISC. For example, the mtDNA replicative helicase in insect lineages is likely to carry a [2Fe-2S] cluster, which is likely not present in mammals (or perhaps any vertebrate species), and this enzyme is absent in yeast, despite its evolutionary relatedness to the primase-helicase of bacteriophage T7. To date, no ISC-containing protein in mitochondrial RNA metabolism has been identified, and given the structural and catalytic proper-

ties of proteins involved in mitochondrial transcription, RNA processing and storage, and translation, this is surprising. Thus, there remains much to explore in the realm of ISC biology within the mitochondrion.

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